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those all needs as soon
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in covalent bonding
with there is an overlap
of parallel orbitals - this
type of attachment
occurs pi bond In what
form do electrons such
as hydrogen, nitrogen

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and oxygen normally
occur

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Covalent Bonding
Single Covalent Bonds
When only one pair of
electrons is shared,
such as in a hydrogen
molecule, it is a single
covalent bond. The
shared electron pair is
often referred to as the

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bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally attracts the pair of shared electrons.

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Single Covalent Bonds
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electrons is shared,

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such as in a hydrogen molecule, it is a single covalent bond. The shared electron pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally attracts the pair of shared electrons.

Chapter 8: Covalent Bonding

Chapter 8 Covalent Bonding and Molecular

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Bonding Answer

Structure 8-11. nuclei. This results in stronger attractive forces between electrons and nuclei, decreasing the distance between the nuclei. A carbon-carbon single bond has a bond order of 1 and is longer than a carbon-carbon double bond with a bond order of 2.

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Chapter 8 121 Section

8.1 The Covalent Bond

pages 240–247

Practice Problems page

244 Draw the Lewis

structure for each

molecule. 1. PH₃ H₂ O₂

H₂ H₂ P₂ respectively,

for single, double, and

triple P₂ — — 2. H₂ S₂ H₂

H₂ H₂ — H₂ S₂ S₂ ...

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More electronegative atoms attract electrons more strongly and gain a slightly negative charge in their bonds.

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Covalent bonding occurs between two non-metallic atoms characterized by the sharing of electron pairs between the atoms and other covalent bonds with electronegativity difference is greater than 2.0 (<2.0). In the case of covalent bond formation, polyatomic

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ions are formed.

Covalent Bond - Definition, Types, Properties, and Examples

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Covalent Bonding

Single Covalent Bonds

When only one pair of electrons is shared, such as in a hydrogen molecule, it is a single covalent bond. The shared electron pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 84, each covalently bonded atom equally attracts the pair of shared

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8.2 The Nature of
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Experimental evidence,
however, indicates ...

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correct name for AlBr_3
?

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.... The chemical bond
that results from
sharing valence
electrons is a covalent
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Distinguish between
the terms
electronegativity
versus electron affinity,
covalent bond versus
ionic bond, and pure
covalent bond versus

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polar covalent bond.

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