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Answer:  $4.93 \times 10^{-5}$  L or 49.3  $\mu$ L In Example 12.2.1 and Example 12.2.2, the identity of the limiting reactant has been apparent: [Au(CN) 2] <sup>-</sup>, LaCl 3, ethanol, and para-nitrophenol. When the limiting reactant is not apparent, we can determine which reactant is limiting by comparing the molar amounts of the reactants with their ...

### Chapter 12.2: Stoichiometry of Reactions in Solution ...

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